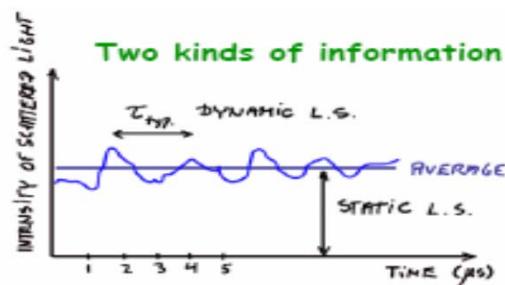
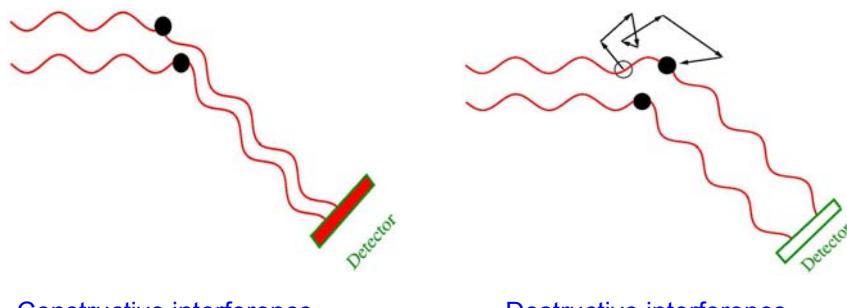


## Static light scattering vs. Dynamic light scattering

- Static light scattering measures time-average intensities (mean square fluctuations)  
molecular weight  
radius of gyration  
second virial coefficient
- Dynamic light scattering measures real-time intensities  $i(t)$ , and thus dynamic properties  
diffusion coefficient  
(hydrodynamic radius)  
size distributions



## How DLS Works: Interference of Light



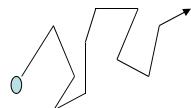
Brownian motion results in light intensities which fluctuate in time.

The timescale of those intensity changes is a measure of the diffusion constant.

Brownian motion is a phenomenon that is fundamental to this experiment.

It describes the way in which very small particles move in fluid suspension, where the fluid consists of molecules much smaller than the suspended particles.

The motion of suspended particles is random in nature, and arises from the cumulative effect of bombardment by the suspending medium's molecules.

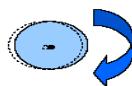


Molecules in a liquid are constantly in motion, randomly bouncing off one another. As these molecules move around in the liquid, they are also bouncing off any suspended particles in a random manner, imparting a momentum to the suspended particles, the magnitude and direction of which fluctuate in time.

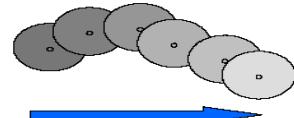
It is the resulting 'random walk' behavior of the suspended particles that is called Brownian motion, and that randomness makes this experiment possible.

#### Diffusion of molecules ---- Brownian Motion

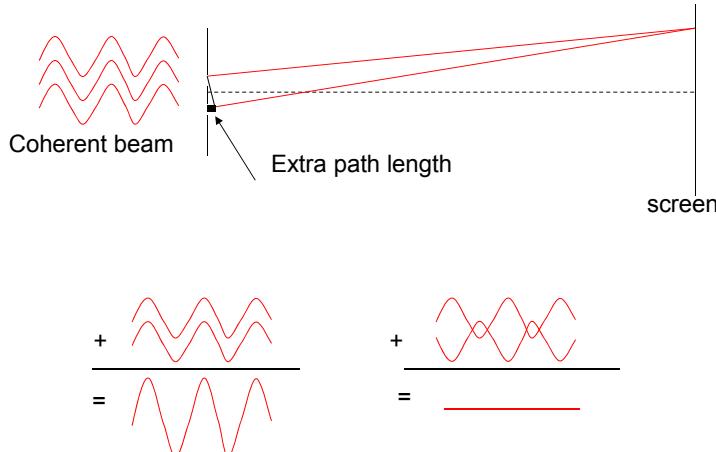
Rotational diffusion: no signal change



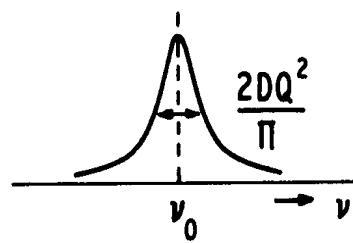
Translational diffusion: signal change



## Double Slit Experiment



- The incident laser light is highly monochromatic ( $\nu_0$ )
- The scattered radiation will display a range of frequencies, due to the Doppler shift caused by Brownian motion.
- For particles moving away from the detector,  $\nu < \nu_0$
- For particles moving toward the detector,  $\nu > \nu_0$
- The technique is also called “quasi-elastic light scattering”, reflecting the apparent change in frequency of the scattered radiation.
- The spread of frequencies is dependent on the diffusion coefficient  $D$ .

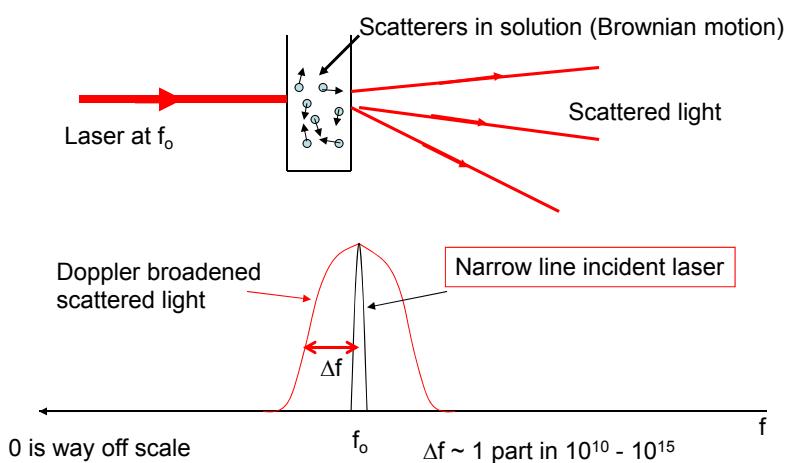


$$\Delta\nu_{1/2} = \frac{2Dq^2}{\pi} \text{ where } \Delta\nu_{1/2} \text{ is the frequency}$$

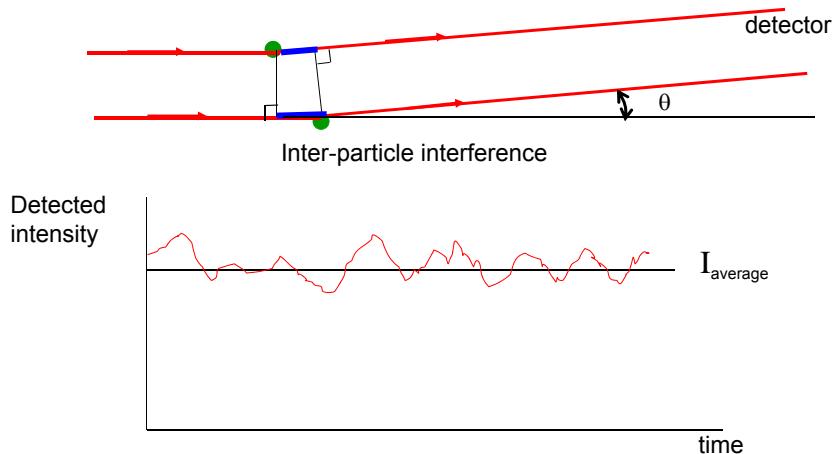
$$\text{spread at half - height and } q = \frac{4\pi}{\lambda} \sin\left(\frac{\theta}{2}\right)$$

The line width at half height of the spectrum of frequencies observed is related to the diffusion coefficient of the scatterer

### Light Scattering Experiment



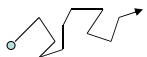
### More Detailed Picture



How can we analyze the fluctuations in intensity?

$\text{Data} = g(\tau) = \langle I(t) I(t + \tau) \rangle_t$  = intensity autocorrelation function

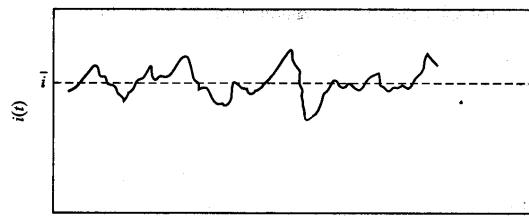
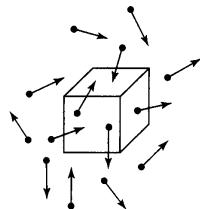
### What determines correlation time?

- Scatterers are diffusing – undergoing Brownian motion – with a mean square displacement given by  $\langle r^2 \rangle = 6D\tau_c$  (Einstein)  

- The correlation time  $\tau_c$  is a measure of the time needed to diffuse a characteristic distance in solution – this distance is defined by the wavelength of light, the scattering angle and the optical properties of the solvent – ranges from 40 to 400 nm in typical systems
- Values of  $\tau_c$  can range from 0.1  $\mu\text{s}$  (small proteins) to days (glasses, gels)

### Dynamic light scattering experiment:

- Light intensity is  $I(t)$  at time  $t$
- At the time  $t + \tau$ , which is a very small time later than  $t$ , the diffusing particles will have new positions and the intensity at the detector will have a value  $I(t+\tau)$
- The detector saves the values for  $I(t + \tau)$  at numerous times
- The autocorrelator automatically calculates the function instead of the discrete intensities

### Correlation spectroscopy



$$\Delta i(t) = i(t) - \bar{i} \text{ where } \bar{i} \text{ is the average scattering intensity.}$$

Measure  $\Delta i(t)$  and  $\Delta i(t + \tau)$

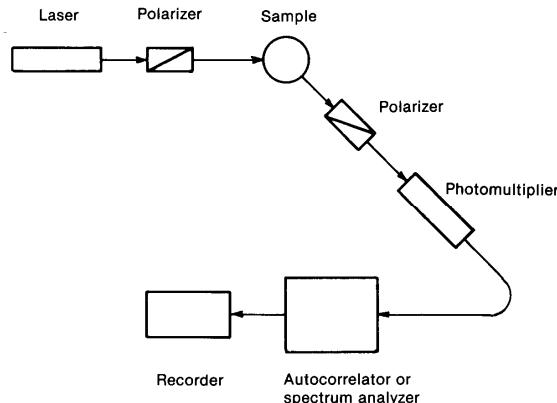
where  $\tau$  is an arbitrary delay time.

Define an autocorrelation function :

$A(\tau) = \overline{\Delta i(t) * \Delta i(t + \tau)}$  where the bar denotes an average over many observations.

## Correlation spectroscopy

The number of macromolecules present in the volume element illuminated by the laser fluctuates with time. Thus, the scattering intensity also fluctuates.



In a single detector setup, autocorrelation takes the place of cross-correlation.

Autocorrelation is mathematically identical to cross-correlation, except that rather than comparing two signals with one another, one signal is compared with a time-delayed version of itself.

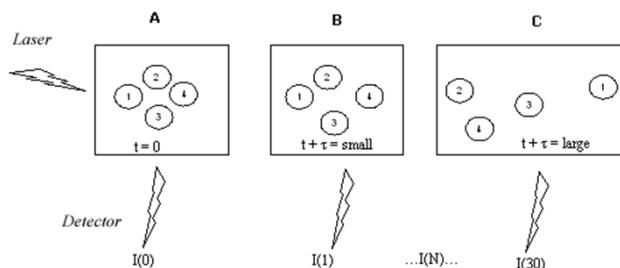
$I(t+\tau)$  correlates with  $I(t)$ , the closer the measurement is to time zero, the more similar  $I(t+\tau)$  is to  $I(t)$  since the particles have not had much time to move:

$$\lim_{\tau \rightarrow 0} I(t + \tau) = I(t)$$

As time goes on there is no more similarity between the starting state and the current state - the measured intensities do no correlate anymore to the beginning one.

This happens faster if the particles are smaller since smaller particles move faster. One needs a method for quantifying how fast the correlation takes to break down between the starting measurement and one a short time later.

- Light intensity is  $I(t)$  at time  $t$
- At  $t + \tau$  new positions of particles - the intensity at the detector will have a value  $I(t+\tau)$
- The detector saves the values for  $I(t + \tau)$  at numerous times
- The autocorrelator automatically calculates the function instead of the discrete intensities



The function used to calculate this correlation is the *autocorrelation function* it describes how a given measurement relates to itself in a time dependent manner.

At time zero there is a 100% autocorrelation.

As time progresses, the autocorrelation diminishes reaching zero as there is no more similarity between starting and ending states.

The decay of the autocorrelation is described by an exponential decay function  $g(\tau)$  which relates the autocorrelation to the diffusion coefficient  $D$  and the measurement vector  $q$ :

$$g(\tau) = g_o \exp[-Dq^2\tau]$$

$$q = \frac{4\pi n}{\lambda} \sin\left(\frac{\theta}{2}\right)$$

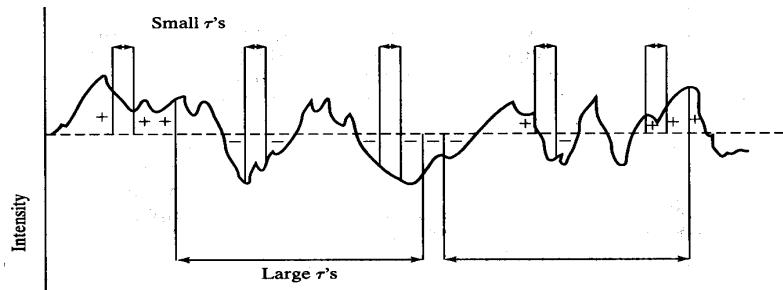
$n$  = refractive index of the solution (1.33 for water)

$\lambda$  = wavelength of the laser (e.g 632.8 nm)

$\theta$  = angle of scattering measurement

The intensity varies in the time scale of the molecules moving the distance of a wavelength:  
ovalbumin (45 kDa,  $D = 8.10^{-11} \text{m}^2/\text{s}$ ) moves 828 nm (laser wavelength) in 4 ms

As indicated in the figure,  $g(\tau)$  will tend to be  $>0$  for short  $\tau$  and will be  $\approx 0$  for long  $\tau$ . Thus, if we measure  $g(\tau)$  as a function of  $\tau$ , we observe an exponential decay to zero.



$$g(\tau) = g_o \exp[-Dq^2\tau]$$

Thus, a plot of  $\ln[g(\tau)]$  vs  $\tau$  will have a slope of  $-Dq^2$ .

### Brownian motion

Considering the 1-D movement of a suspended particle in a solvent:

In the steady state:  $\langle V(t) \rangle = 0$

The statistical average of the product of the velocities at two different times depends only on the time difference:

$$\begin{aligned} \langle V(t_1)V(t_2) \rangle &= C_v(t_1-t_2) = 2D\delta(t_1-t_2) \\ \langle V(t_1)V(t_2) \rangle &= \langle V(t)V(0) \rangle = C_v(t) = 2D\delta(t) \end{aligned}$$

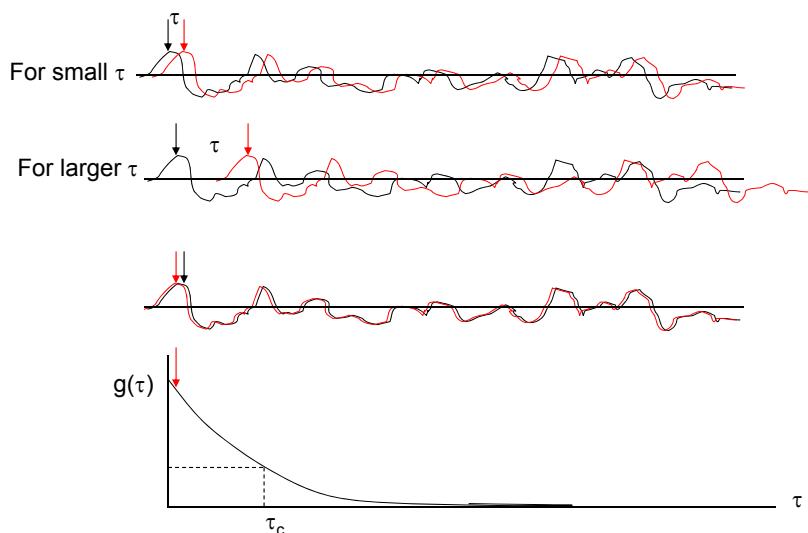
Where, **D** is the diffusion coefficient and  **$C_v(t)$**  is the velocity correlation function that is equivalent to the thermal motion for short time difference

## Autocorrelation Function

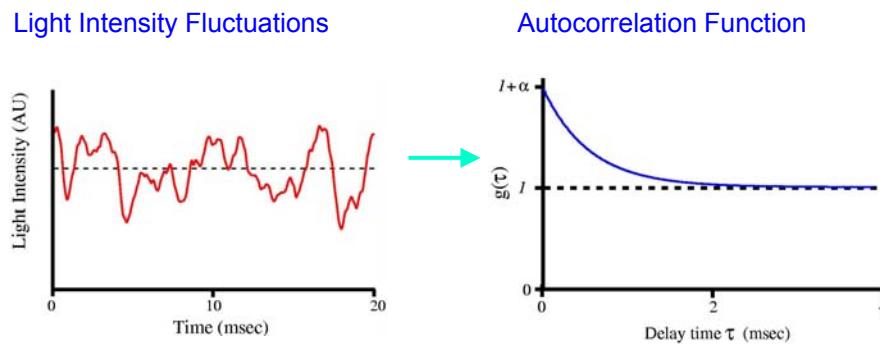
- The average of the time varying portion of the intensity at some initial time,  $t$ , with the time varying portion of the intensity at some later time,  $t+\Delta t$ .
- This calculation is done for many values of  $\Delta t$ .
- For small values of  $\Delta t$ , the signal is still correlated with the signal at  $t = 0$ ; positive average intensity.
- For large values of  $\Delta t$ , this is not true, and the autocorrelation function will eventually average out to zero.
- By measuring how long it takes the function to go to zero, we can tell how fast the particles are moving

### Intensity autocorrelation

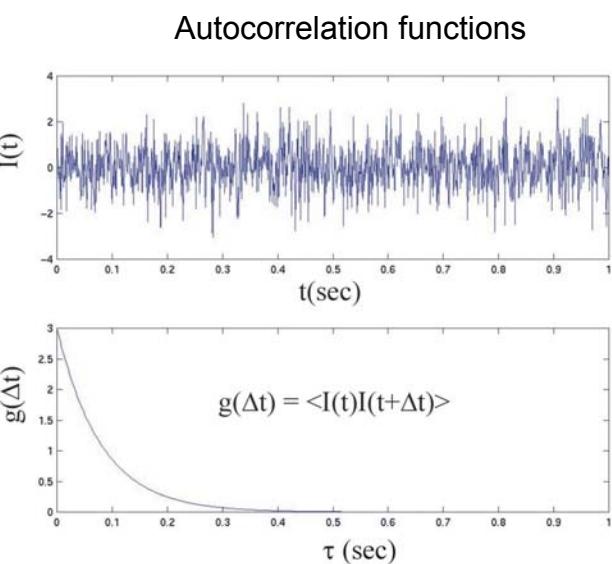
- $g(\tau) = \langle I(t) I(t + \tau) \rangle_t$



## Analysis of Light Intensity Fluctuations



Particles with a large physical dimension (radius) diffuse more slowly through a solvent, while small particles diffuse more quickly. Intensity fluctuations seen through time are therefore slower for large particles.



## Diffusion

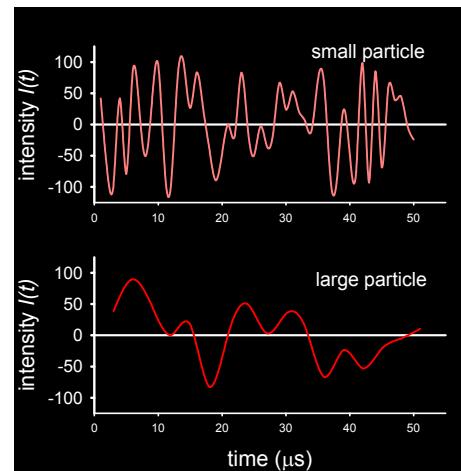
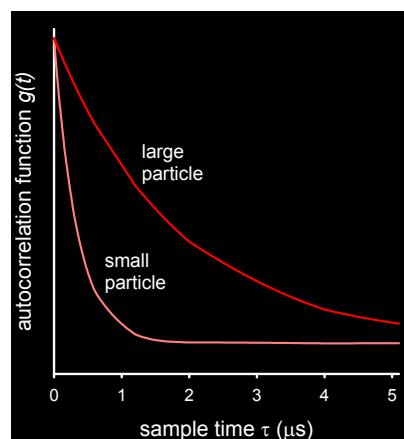
- What can we learn from the correlation time?
- Knowing the characteristic distance and correlation time, we can find the **diffusion coefficient D**
- According to the Stokes-Einstein equation

$$D = \frac{k_B T}{6\pi\eta R}$$

where  $R$  is the radius of the equivalent hydrodynamic sphere and  $\eta$  is the viscosity of the solvent

- So, if  $\eta$  is known we can find  $R$  (or if  $R$  is known we can find  $\eta$ )

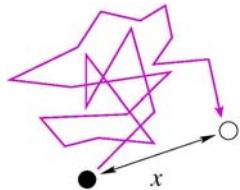
Smaller particles diffuse faster



## Timescale of Motion

How does one get from a diffusion coefficient to  $R_h$ ?

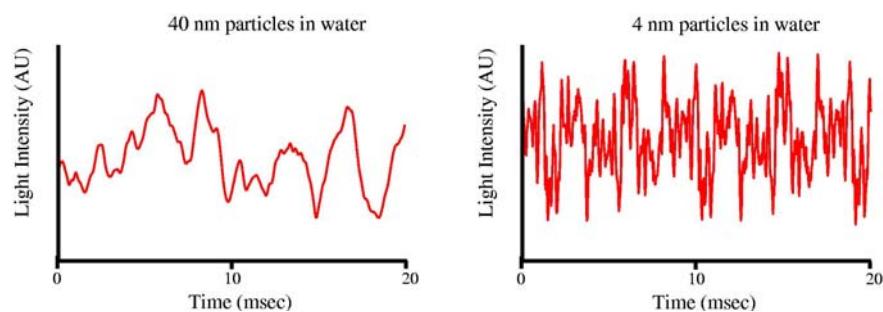
Stokes - Einstein Relation



$$R_h = \frac{k_B T}{6\pi\eta D_t}$$

$k_B$  – Boltzmann's constant  
 $T$  – temperature (Kelvin)  
 $\eta$  – viscosity of solvent  
 $R_h$  – hydrodynamic radius

## Intensity Fluctuations



The rate at which particles diffuse is related to their size, provided all other parameters are constant.

By fitting the points of autocorrelation to the function  $g(\tau)$ , the diffusion coefficient can be measured and related to the equivalent sphere of diameter  $d$  using the Stokes - Einstein equation:

$$D = \frac{k_B T}{3\pi\eta d}$$

$\eta$  = diluent viscosity (water =  $8.94 \times 10^{-4}$  kg/(ms)

T = Temperature (K) (room temp = 298 K)

D = diffusion coefficient (in m<sup>2</sup>/s)

$k_B$  = Boltzmann constant ( $1.3807 \times 10^{-23}$  J/K)

d = sphere diameter (m)

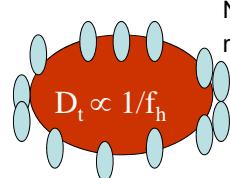
### What Affects Translational Diffusion?

$$D_t \propto 1/d$$

Small particles move faster

$$D_t \propto 1/f_s$$

Nonsphericity slows down mobility



Intrinsic factors

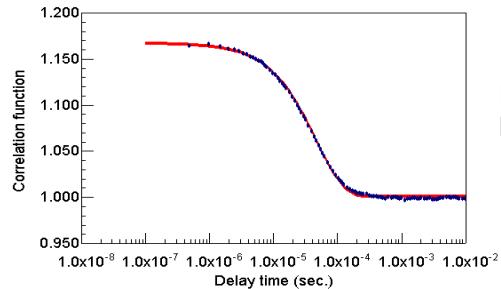
$$D_t \propto 1/\eta$$

Viscous solvent slows it down.  
...and if concentration too high, 'viscosity effects'

$$D_t \propto T$$

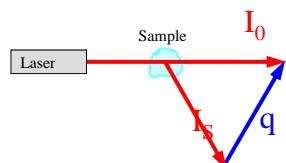
High temperature speeds it up

### Autocorrelation Function

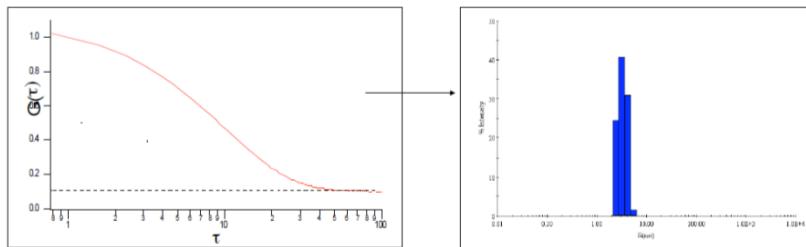


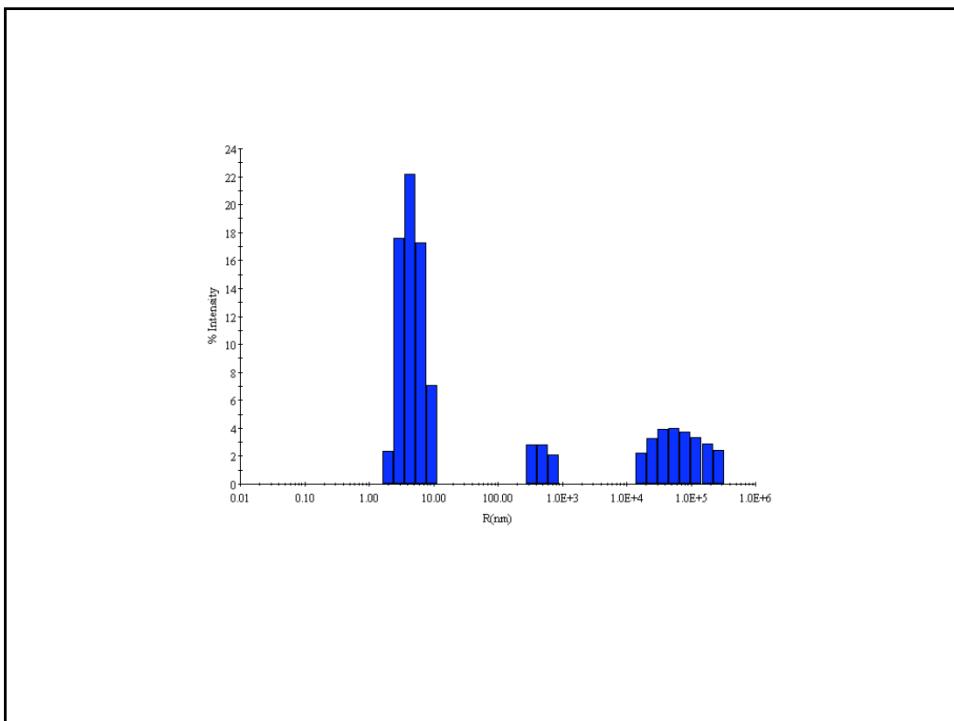
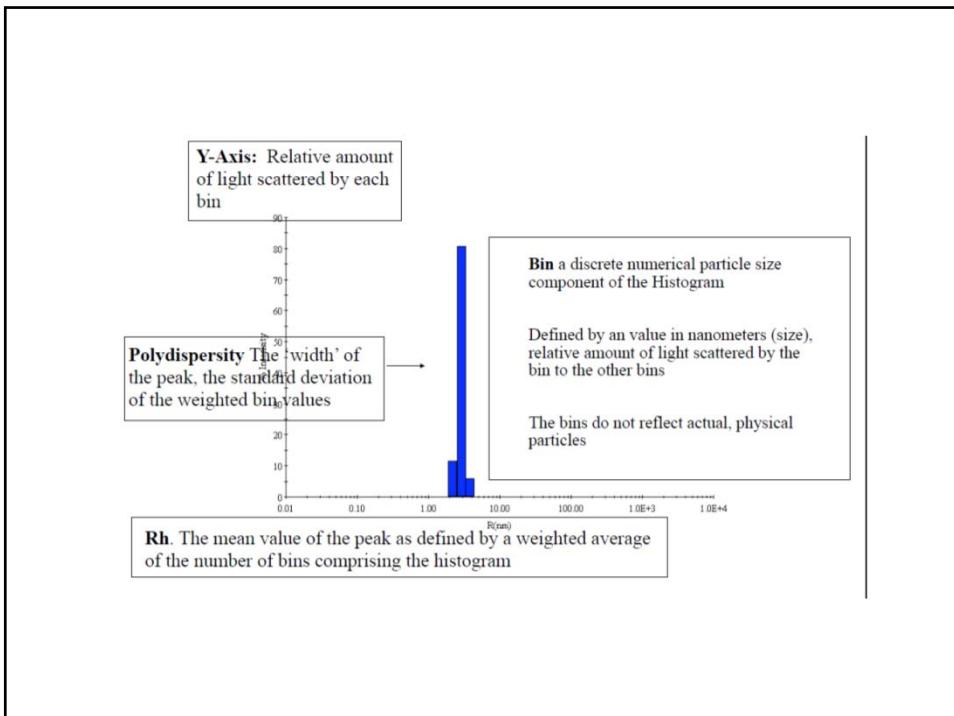
$R_h = 9 \text{ nm}$   
latex spheres

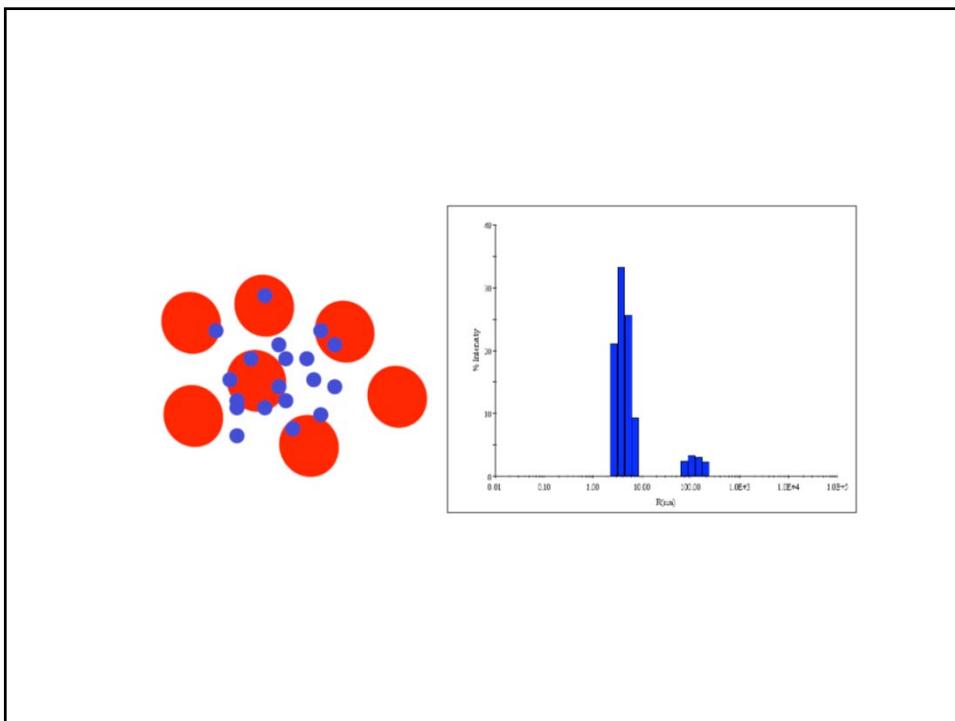
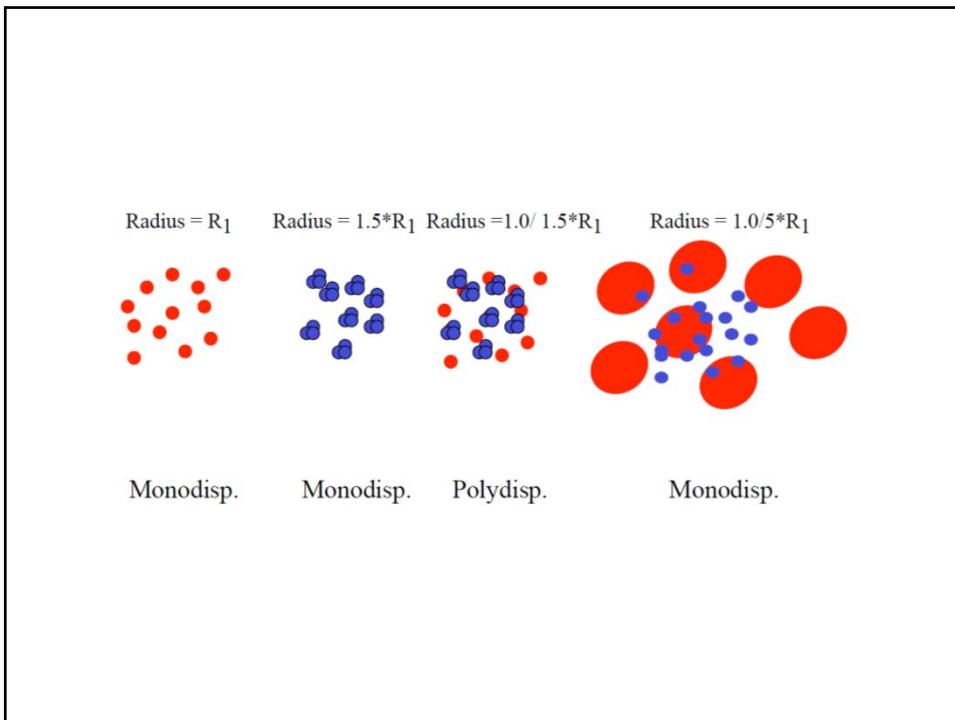
Autocorrelation function:  $g^2(\tau) = \langle I \rangle^2 \left( 1 + \alpha \exp(-2D_t q^2 \tau) \right)$



$$q = \frac{4\pi n_0}{\lambda_0} \sin(\theta/2)$$

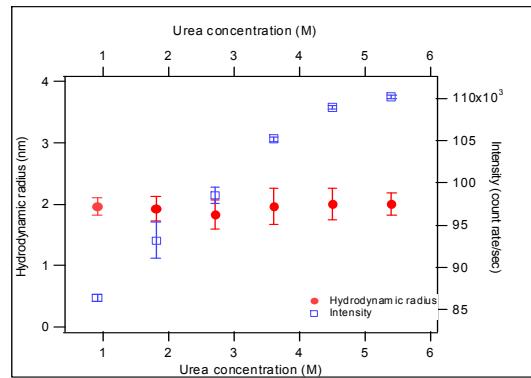






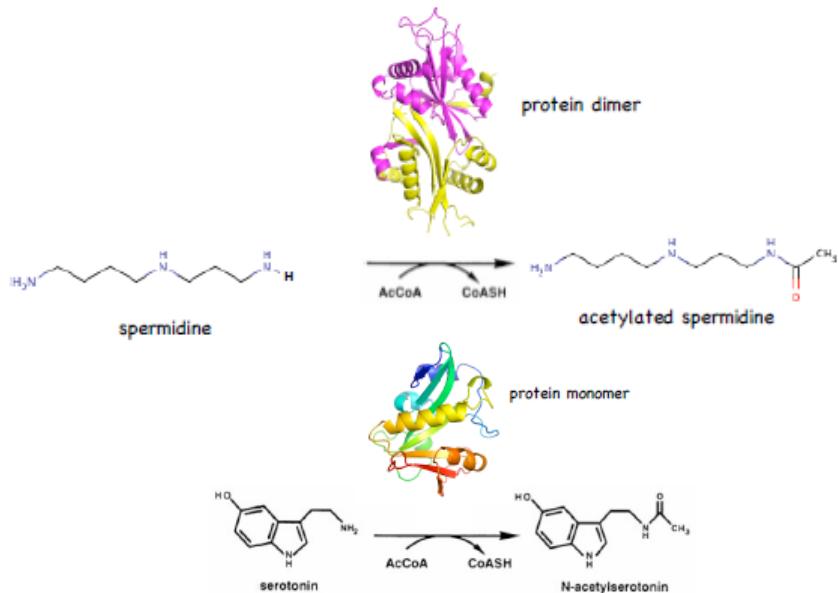
## Hydrodynamic radius of lysozyme as a function of urea concentration

Concentration of urea (M)	Hydrodynamic radius (nm)
0.9	1.96 $\pm$ 0.14
1.8	1.93 $\pm$ 0.20
2.7	1.83 $\pm$ 0.23
3.6	1.96 $\pm$ 0.29
4.5	1.89 $\pm$ 0.25
5.4	1.99 $\pm$ 0.18



- The hydrodynamic radius of lysozyme (1mg/ml) is independent of the urea concentration

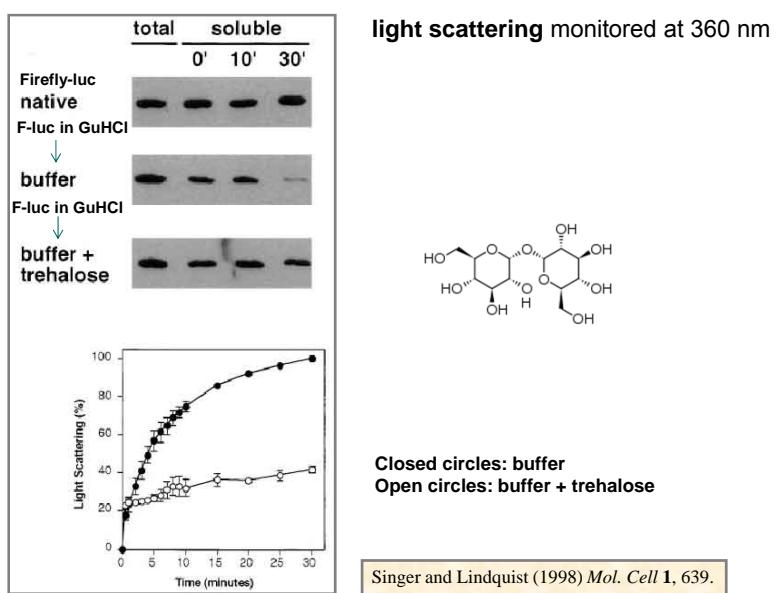
Spermine/spermidine acetyl transferase is similar in a.a. sequence and structure to serotonin acetyl transferase, but it is a dimer, with two subunits of about 170 a.a. each.

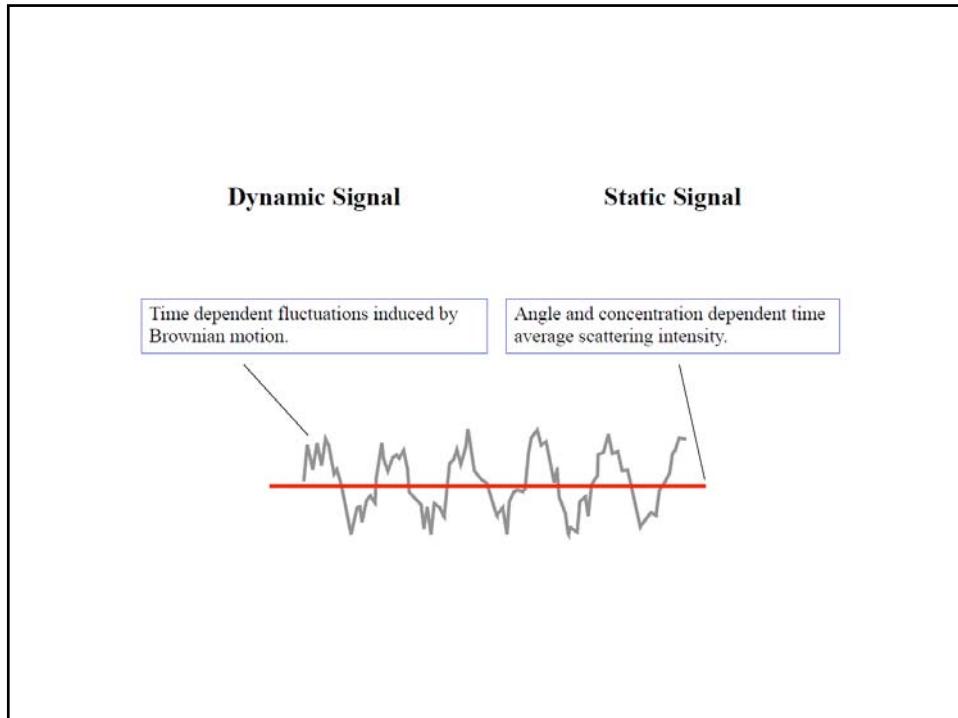


The data given below describe the variation of sedimentation coefficient,  $s$ , and diffusion coefficient,  $D$  for a protein as a function of pH.

pH	2	3	4	5	6	7	8	9	10
s (S)	2.93	3.02	3.89	4.41	4.40	4.15	3.60	2.25	2.20
D ( $\times 10^7$ ) $\text{cm}^2\text{s}^{-1}$	7.91	8.00	8.00	5.90	5.92	5.61	4.86	3.08	2.97

Explain what happens to the protein at low and high pH, assuming that the protein adopts its native structural state (i.e. structural conformation) between pH 5 and pH 6.





<u><b>Dynamic light scattering</b></u> real-time intensities	<u><b>Static light scattering</b></u> time-average intensities
<ul style="list-style-type: none"> <li>• Diffusion coefficient <math>D_T</math></li> <li>• Modality</li> <li>• Hydrodynamic radius <math>R_h</math></li> <li>• Size Distribution</li> <li>• Estimation of MW</li> </ul>	<ul style="list-style-type: none"> <li>• Molecular weight</li> <li>• Radius of gyration</li> <li>• Second virial coefficient</li> </ul>